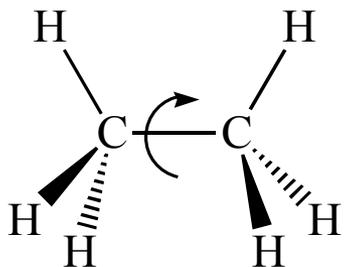
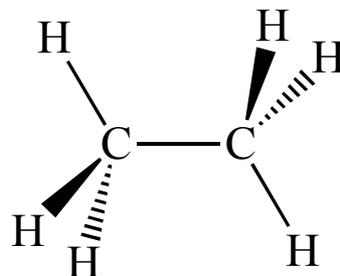


Conformers – Ethane

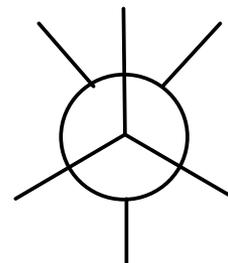
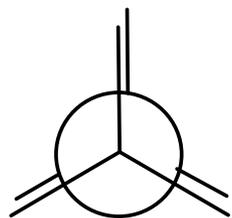
One important aspect of this bond's symmetry is that it is possible to rotate one end of the molecule with respect to the other leading to different molecular *conformations*.



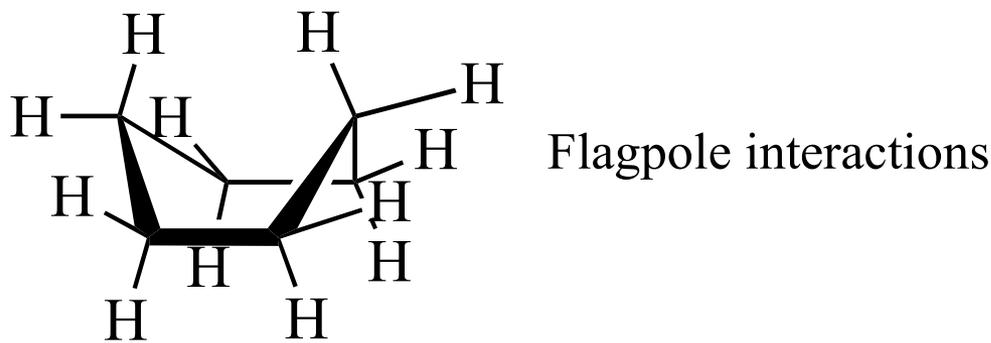
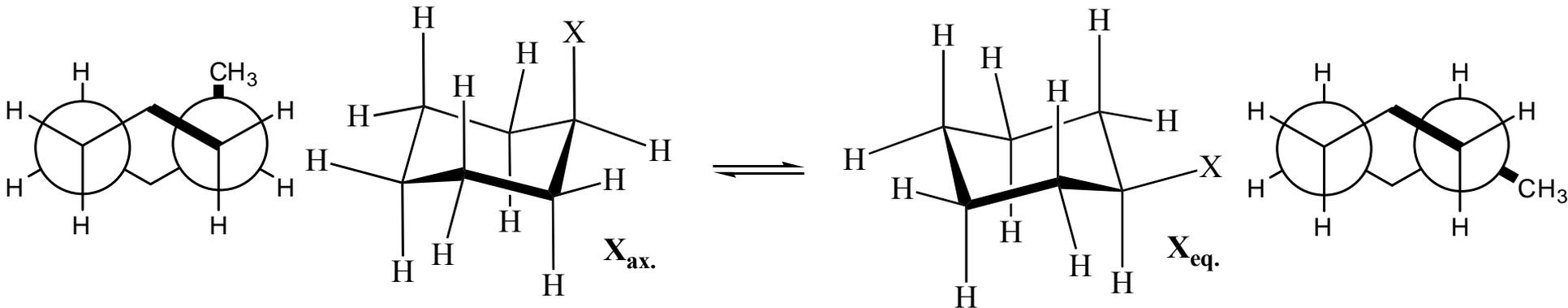
Eclipsed



Staggered



Newman Projections

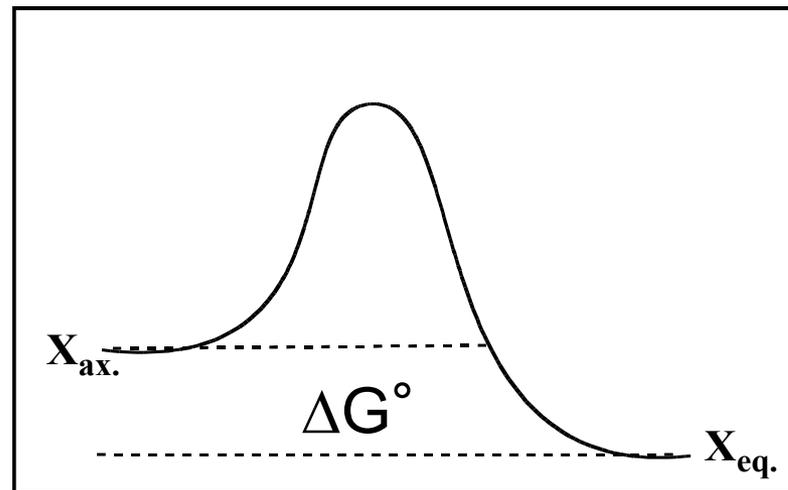


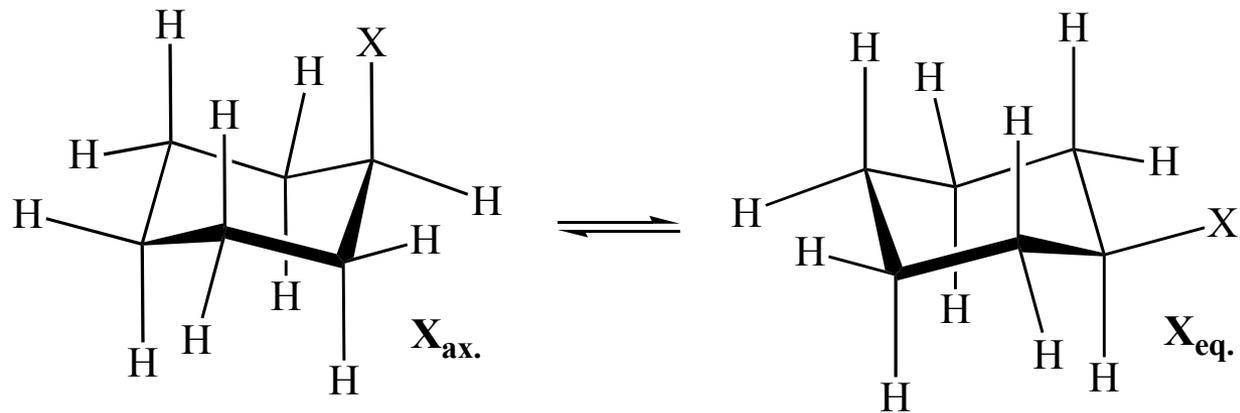
$$K_{eq} = \frac{[X_{eq}]}{[X_{ax}]}$$

$$\Delta G^\circ = -RT \ln K_{eq}$$

Define the "A value" as $-\Delta G^\circ$

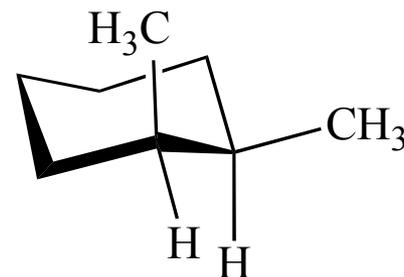
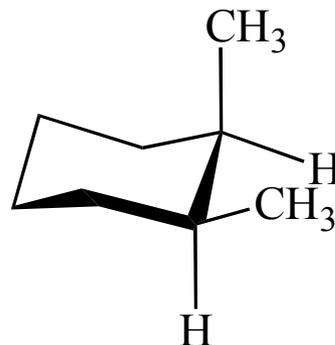
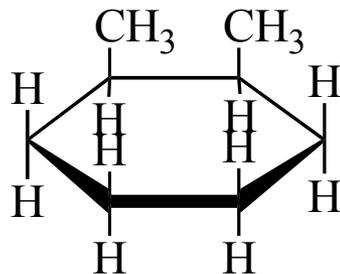
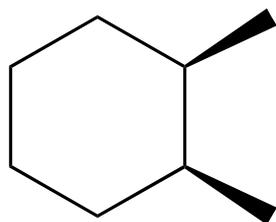
Energy ↑



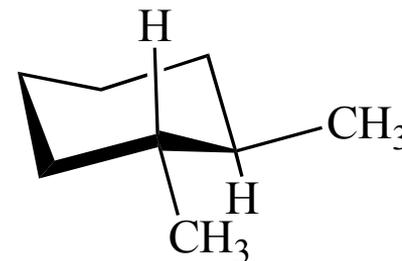
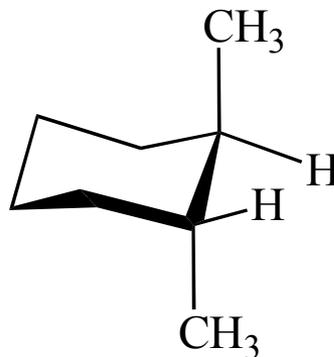
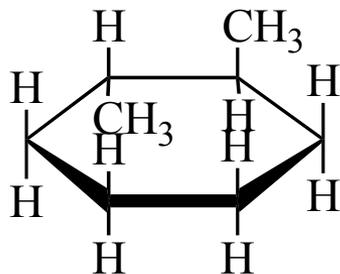
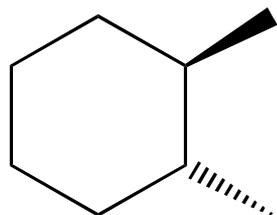


X	K_{eq}	ΔG° (kcal/mol)
H	1	0
CH ₃	19	-1.74
CH ₂ CH ₃	21	-1.79
C(CH ₃) ₃	3300	-4.8
—CH=CH ₂	15	-1.6
—C≡CH	2.2	-0.46
F	1.7	-0.3
Cl	3	-0.6
Br	3	-0.6
I	2.3	-0.5

Conformers – Disubstituted Cyclohexanes

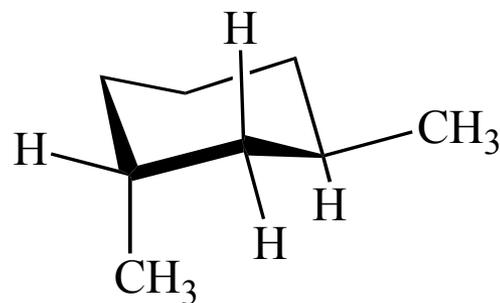
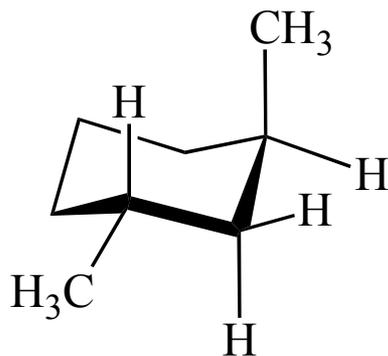
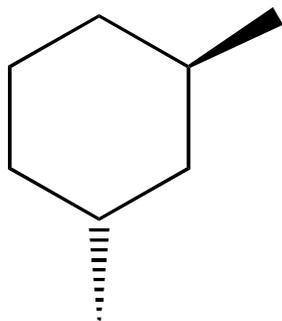
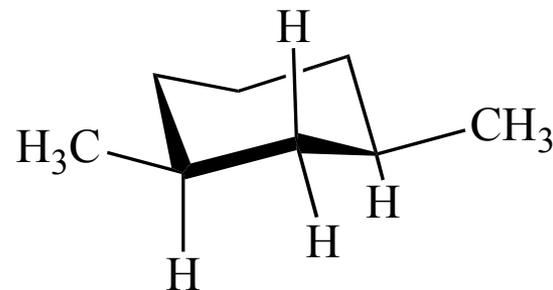
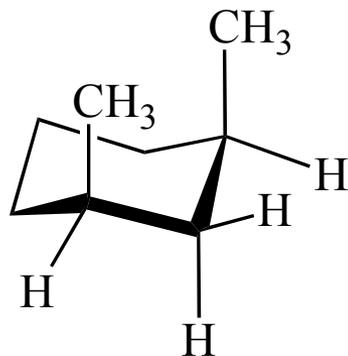
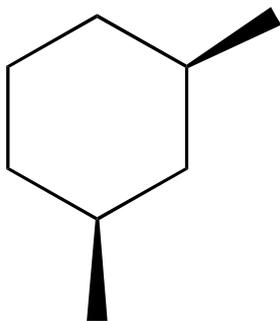


cis-1,2-dimethylcyclohexane

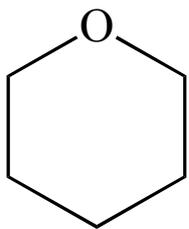
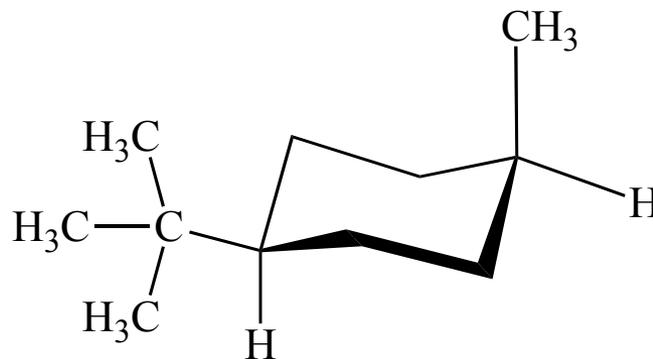
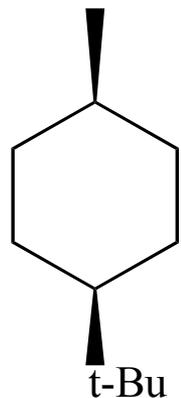


trans-1,2-dimethylcyclohexane

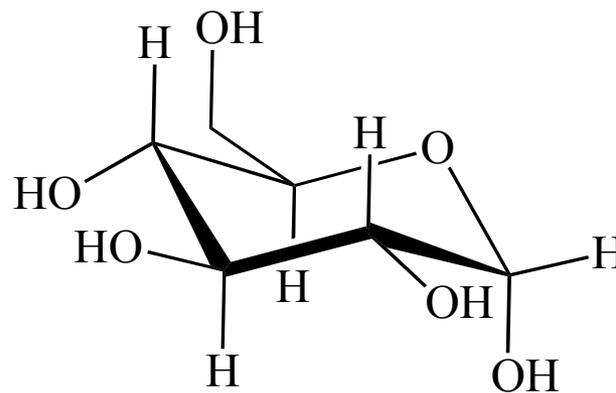
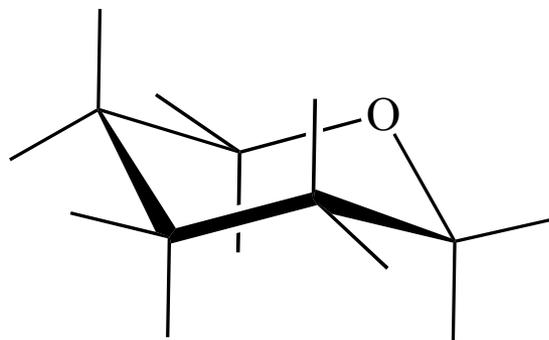
Conformers – Disubstituted Cyclohexanes



Conformers – Disubstituted Cyclohexanes



Tetrahydropyran



α -D-Glucose

Conformers – Cycloalkanes and Chirality

- Remember, ring flips are like bond rotations – they only change the conformation of the molecule. Therefore, you are allowed to do ring flips to determine if your molecule is chiral or not.
- Convince yourself that *cis*-1,2-dichlorocyclohexane is achiral.
- Convince yourself that *trans*-1,2-dichlorocyclohexane is chiral.

Conformers – Smaller Cycloalkanes

- Six-membered rings are prevalent in nature because they are relatively free from:
 - torsional strain: eclipsed bondsand
 - angle strain: bond angles that deviate from those predicted by VSEPR theory
- Build cyclopentane, cyclobutane, and cyclopropane. Is there torsional strain? Angle strain? What do you think would be the most stable conformer for each of these rings? Use the amount of strain to rank these rings from most to least stable.